

Sarojini Naidu Vanita Maha Vidyalaya

Best Practices by Departments

1. Department of Botany

Best practices & Activities Undertaken:

1. Maintaining eco-club.
2. Maintaining Hydroponic Unit.
3. Project work on various topics such as algal member studies on various water samples, identification of plants through ICT tool plant-net and QR code, assessment of water quality from various lakes of the city, Ethno-medicinal studies, assessment of noise pollution through noise level meter at various places in twin cities.
4. Workshop on grafting.

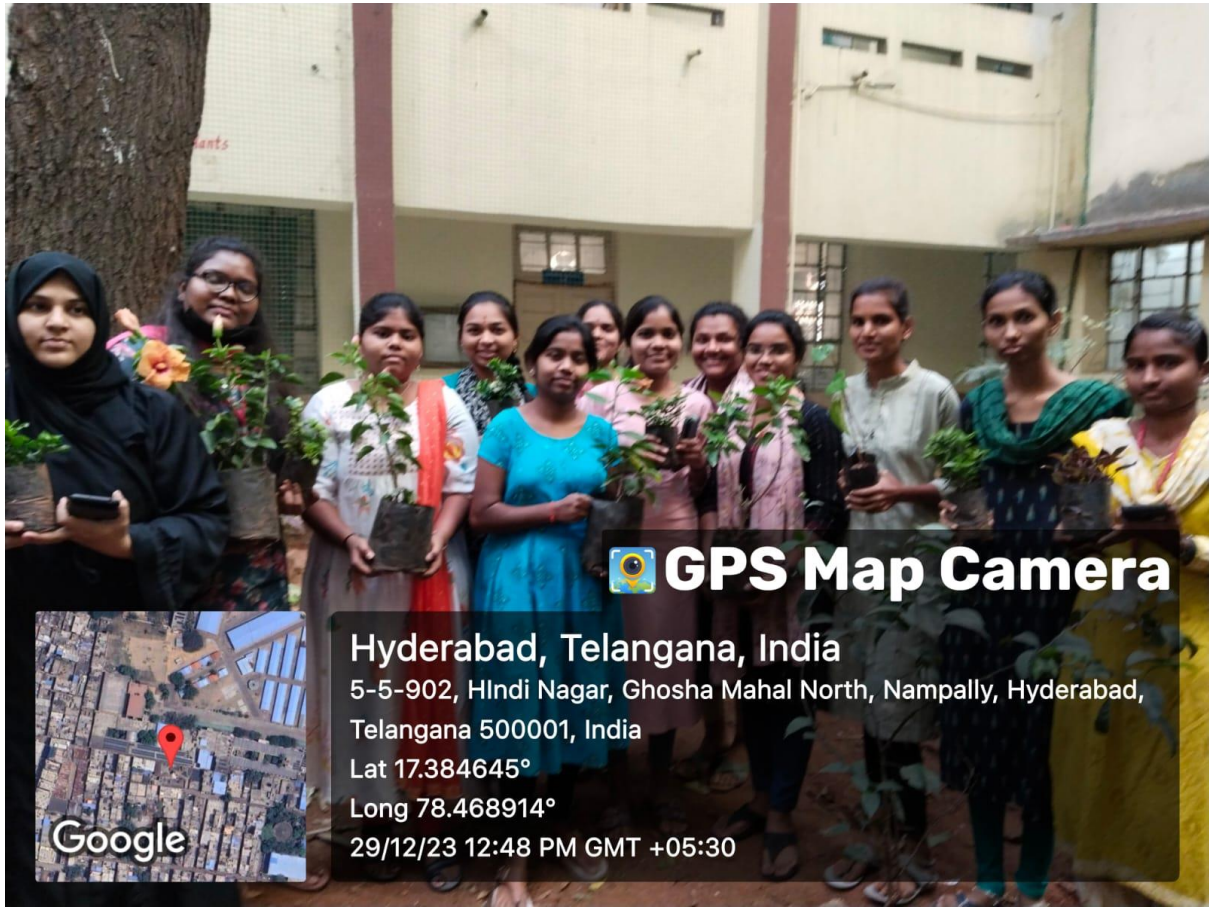
Expected outcome:

1. Students will be sensitized to growing different plants and to conserving, and hand-on experience with gardening.
2. Learning new technique of raising soilless plants.
3. Learning various latest e-technologies in learning subject through various apps available like plantnet, QR Code, using instruments estimation of noise levels at various places, and assessing water quality, and ethnomedicinal studies.
4. Will learn horticultural techniques like grafting and gardening tips from the experts.

Supporting Documents and Photographs:

Eco-Club Activity:






 **GPS Map Camera**

Hyderabad, Telangana, India
5-5-902, Hindi Nagar, Ghosha Mahal North, Nampally, Hyderabad,
Telangana 500001, India
Lat 17.384645°
Long 78.468914°
29/12/23 12:48 PM GMT +05:30



 **GPS Map Camera**

Hyderabad, Telangana, India
5-5-823, Hindi Nagar, Ghosha Mahal North, Nampally,
Hyderabad, Telangana 500001, India
Lat 17.384534°
Long 78.469594°
20/09/22 11:49 AM GMT +05:30







..

Hydroponic Unit

Workshop organized for MSc. students with Prof. Gangadhar on practical knowledge on “Grafting techniques” on February 2019.



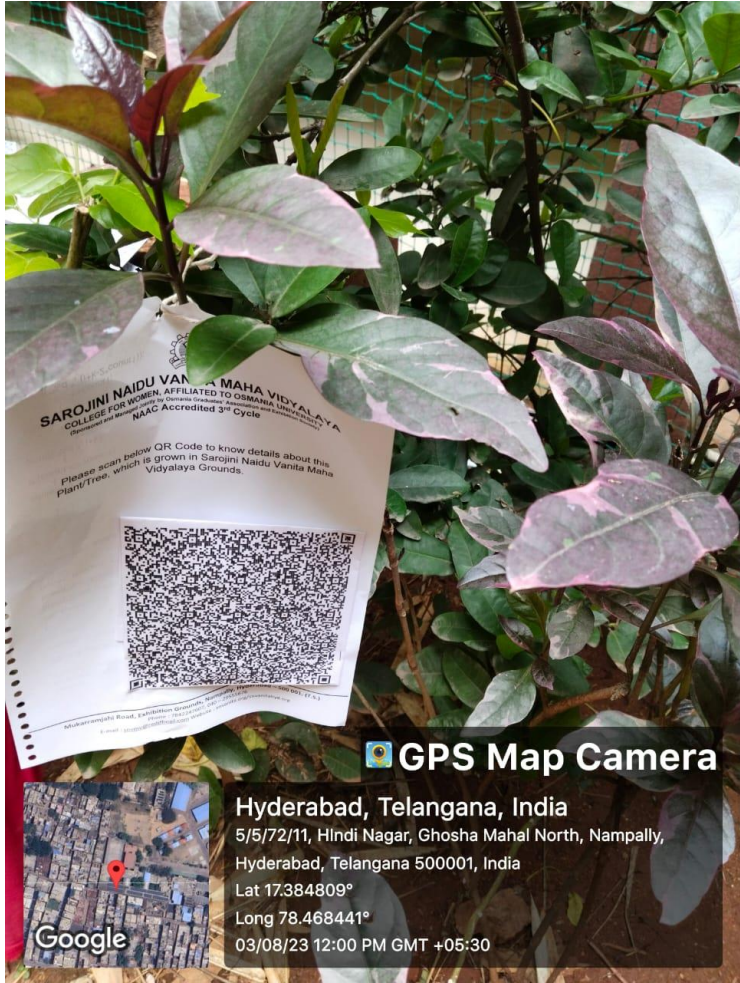
Guest Lecture on grafting techniques



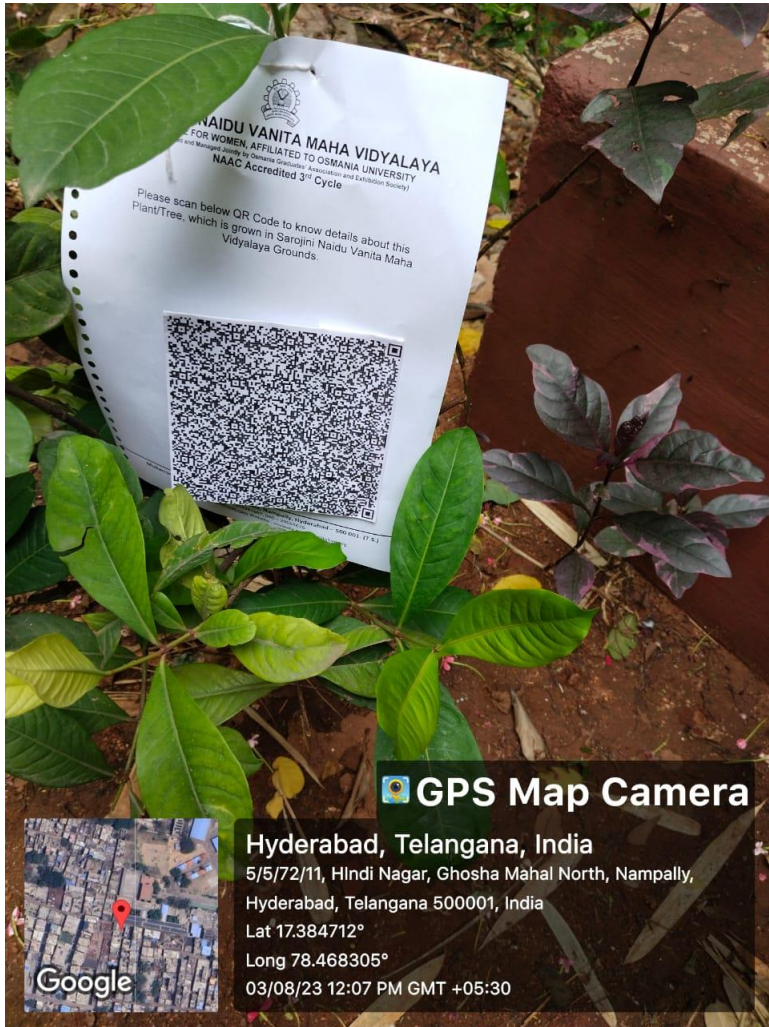
Guest lecture by Prof Gangadhar



Guest lecture on grafting techniques by Prof Gangadhar



Identification of Plants through QR code (mobile app)



Identification of Plants through Plantnet(Mobile app)

12:37

4G+ 75



Guest

Jul 15, 2023 - World flora



Tamarindus indica L.

Tamarind

Fabaceae



Additional data

No GPS location

+ ADD A PLACE OR A COMMENT

Change flora: World flora ▾

Share

Later



12:42

4G

74



Supriya K

Jul 15, 2023 - World flora



Ruellia tuberosa L.

Minnieroot

Acanthaceae



Additional data

No GPS location

+ ADD A PLACE OR A COMMENT

Change flora: World flora

Share

Later



Noise Pollution at certain areas of Hyderabad

1175-21-502-023
M.Bhavani
Msc-Brr

Noise Pollution

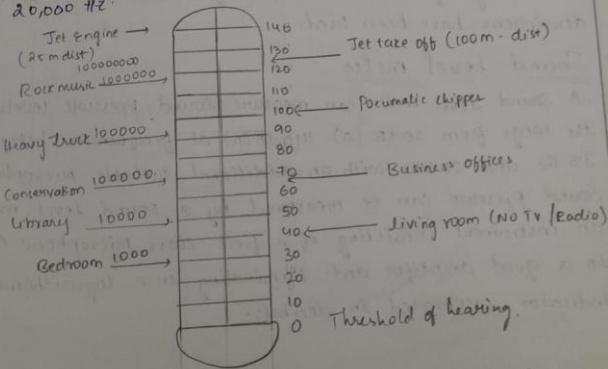
Definition:-

Noise is an unwanted sound energy and is also considered as a pollutant when it exceeds certain limits, noise has a short residence and decay time and hence does not remain in the environment for long period like air or water pollutant.

Nature of a sound:- Sound pressure is also mentioned in newtons per square (N/m²)

- Sound is often described in terms of loudness but this cannot be measured like for sound frequency, sound pressure, and sound intensity.
- Loudness may be described as a distinct auditory impression which is not synonymous with sound pressure.
- Loudness is expressed in a relative unit, called 'sone' one sone equals to the loudness of 40dB sound pressure at 1000 cycles per second.
- Normally hearing is most acute in the frequency range 2000-5500 Hz. Perfect hearing lies in the range of about 16 to 20,000 Hz.

Sound pressure level



Scale showing Common Sound pressure level.

Total Hardness
at given sample of calcium and metals

→ The noise value of sound waves depends upon the following important factors:

- 1) The frequency of sound waves
- 2) The intensity of sound waves
- 3) The time of exposure of sound waves
- 4) Simultaneity of sound waves.

→ The unit of loudness expressed by the mathematical exponent of number 10 is called a bel one tenth of a bel is given for qualitative measurements.

→ Thus a decibel is considered to be the threshold of hearing. The loudness of a sound that a person can withstand without discomfort is about 30db. Automobile horns may reach more than 90 decibels and a jet airplane at a distance of 100 feet may have an intensity of about 140 decibels.

⇒ Measurement of sound :-
Sound energy travels in waves and is measured in frequency and amplitude depending upon the purpose of the wide range of measurement techniques and sound level descriptions have been used.

⇒ Sound level meter :-
A sound level meter can measure sound pressure levels in the range from 20db(A) - 140db(A) at frequencies between 3.5 Hz and 20 kHz with an additional suitable microphone. Sound pressure can be measured by a sound level meter an instrument consisting of a first class microphone coupled to a good amplifier and terminating in a logarithmic indicator calibrated in decibels.

Total Hardness of water
at given sample of water and calcium and metals

⇒ Control of Noise Pollution :-

1. Turn off appliances at home and office.
2. Shut the door when using noisy machines.
3. Use ear plugs.
4. Lower the volume when listening to song, radio and TV.
5. Stay away from noisy areas like industries, air ports, ports.
6. Follow the limits of noise levels.

Go green by planting trees because plants are good noise absorbers. Plants can reduce sound by 5-10 decibels around them.

⇒ Sound recorded in some places are given below :-

S.No	Areas	Sound recorded (db)
1	LB Nagar	92.7 dB
2	Ajzal Gunj	88.2 dB
3	Mata Kept	93.3 dB
4	Musharambagh	80.3 dB
5	Nagole	93.2 dB

R. Kalyani
T. Sudha Rani
Head
Dept. of Botany
Sarojini Naidu Vanita Maha Vidyalaya

Estimation of Total Hardness of water:

M. Bhavani
1175-21-502-023
Msc - Bm

Estimation of Total Hardness of water

⇒ Aim:- To estimate the hardness of given sample of water.

⇒ Principle:- Hardness is generally caused by calcium and magnesium ions in the water. Cations of some other metals like zinc, magnesium are capable to precipitate the soap thus contributing to the hardness. The concentration of these ions is very low in natural water. Calcium & magnesium ion forms a complex of wine red colour has got a stronger affinity towards calcium and magnesium ions and therefore by addition of EDTA [Ethylene diamine Tetraacetic acid], the former complex is broken down and a new complex of blue colour is formed.

⇒ Materials:- laboratory glass ware.

⇒ Reagents:-

1. Ammonium buffer solution - Dissolve 10.5 gms of NH_4Cl in 14 ml of NH_4OH & add distilled water to make up to the volume of 200 ml.
2. EDTA solution - Dissolve 3.7 gms of disodium salt of EDTA in distilled water to prepare a litre of solution.
3. Eriochrome Black 'T' indicator - Dissolve 0.5 gms of EDTA / Eriochrome black 'T' indicator in 100 ml of 80% of Ethyl alcohol.
4. Sodium sulphide solution - Dissolve 5 gms of $\text{Na}_2\text{S} \cdot 9\text{H}_2\text{O}$ in 100 ml of distilled water and kept in the light bottle to avoid oxidation.

⇒ Procedure:-

1. Take 50 ml of sample in a conical flask.
2. Add 1 ml of buffer solution and 1 ml of Na_2S solution.
3. Add 4-5 drops of Eriochrome black 'T' indicator (or) 100 mg EBT powder and shake the solution, it turns to wine red.
4. Titrate the contents against EDTA solution until the wine red colour of the solution turns to blue colour.

→ Inference:- The total hardness is generally caused due to the presence of excess calcium and magnesium ions in the form of carbonates, chlorides and sulphates and they impart permanent hardness to water.

→ Tap water contains 240 mg/lit of hardness
 → The sample of water shows the high concentration of Ca & Mg ions which in long term use causes joint pains, brittling of bones and teeth.

- Total Hardness of water in Tapwater is 240 mg/lit
- Total Hardness of water in Pondwater is 60 mg/lit

→ Hardness of water in Tapwater:

SNO	Initial	Final	Vol of EDTA
1	0	12 ml	12 ml
2	0	14 ml	14 ml
3	0	14 ml	14 ml

Total hardness of water mg/lit
 $= \frac{\text{ml of EDTA used}}{\text{ml of water sample}} \times 10$
 $= \frac{12}{50} \times 100$
 $= 240 \text{ mg/lit}$

Hardness of water in Pond water:

SNO	Initial	Final	Vol of EDTA
1	0	3.5 ml	3.5 ml
2	0	3 ml	3 ml
3	0	3 ml	3 ml

Hardness of water mg/lit = $\frac{\text{ml of EDTA used}}{\text{ml of water sample}} \times 10$
 $= \frac{3}{50} \times 1000$
 $= 60 \text{ mg/lit}$

R. Kalyani

Result :-

1. Hardness of calcium in given Pond water is 80 mg/Ltr
2. Hardness of calcium in given Tap water is 480 mg/Ltr

⇒ Hardness of calcium in Tap water

SNO	Initial	Final	Vol of EDTA
1	0	24ml	24ml
2	0	24ml	24ml
3	0	24ml	24ml

Hardness of calcium in Tap water = $\frac{\text{ml of EDTA}}{\text{ml of sample}} \times 1000$
 $= \frac{24}{50} \times 1000$
 $= 0.48 \times 1000$
 $= \boxed{480 \text{ mg/Ltr}}$

⇒ Hardness of calcium in Pond water

SNO	Initial	Final	Vol of EDTA
1	0	4ml	4ml
2	0	4ml	3.5ml
3	0	4ml	4ml

Hardness of calcium in Pond water = $\frac{\text{ml of EDTA}}{\text{ml of sample}} \times 1000$
 $= \frac{4}{50} \times 1000$
 $= 0.08 \times 1000$
 $= \boxed{80 \text{ mg/Ltr}}$

T. Swalva Rani
 Head
 Dept. of Botany
 Sri Lalit Mahila Vaidya Vidyalaya

R. Kalyan

Estimation of calcium in the given water sample:

Estimation of Total Hardness of calcium in the water

- Aim :- To estimate the total hardness of calcium (Ca^{2+}) in the given water sample.
- Principle :- Many complex forming ions or molecules leads to a step wise formation of complexes, each complex is characterised by its own stability constant. This property of step wise formation of complex can be used in titrimetry and one can be able to estimate the concentration of the respective ions usually EDTA and Aminofoic polycarboxylic acid is a good complexing agent is used for such estimation. In this titration metallochromic indicators [Solo chrome dark blue for Calcium ions is used]. The reactions are Calcium ions, Buffer solution $\text{pH} = 10$, then free ions of calcium 2 electrons rise to the Calcium ions



Indicator dye [Solochrome] + $2e^- \longrightarrow$ Indicator metal complex
Indicator metal complex (red coloured complex)

EDTA Solution \longrightarrow metal-EDTA complex + indicator.

⇒ Materials :- Burette, Pipette, Test tube and conical flask.

⇒ Reagents :-

1. NaOH solution - Dissolve 40gms of NaOH in distilled water to make the volume upto 1 litre
2. Murexide indicator - Take 0.2gms of Murexide which is also called as Ammonium purpurate and mix it with 40gms of Potassium Sulphate and this indicator is not stored as it gets oxidised
3. EDTA solution - Dissolve 3.7gms of disodium salt of EDTA in distilled water to prepare a litre of solution.

⇒ Procedure :-

1. Take 50ml of sample in a conical flask and add 1ml of buffer solution and add 4-5 drops of Murexide indicator.
2. Titrate the solution against the EDTA solution until the pink colour of solution turns into purple violet.

⇒ Result :-

1. Hardness of calcium in given Pond water is 80 mg/lit
2. Hardness of calcium in given Tap water is 480 mg/lit

⇒ Hardness of calcium in Tap water

SNO	Initial	Final	Vol of EDTA
1	0	24ml	24ml
2	0	24ml	24ml
3	0	24ml	24ml

$$\begin{aligned}\text{Hardness of calcium in Tap water} &= \frac{\text{ml of EDTA}}{\text{ml of sample}} \times 1000 \\ &= \frac{24 \times 1000}{50} \\ &= 0.48 \times 1000 \\ &= \boxed{480 \text{ mg/lit}}\end{aligned}$$

⇒ Hardness of calcium in Pond water

SNO	Initial	Final	Vol of EDTA
1	0	4ml	4ml
2	0	4ml	3.5ml
3	0	4ml	4ml

$$\begin{aligned}\text{Hardness of calcium in Pond water} &= \frac{\text{ml of EDTA}}{\text{ml of sample}} \times 1000 \\ &= \frac{4}{50} \times 1000 \\ &= 0.08 \times 1000 \\ &= \boxed{80 \text{ mg/lit}}\end{aligned}$$

T. Sudha Rani
Head
Dept. of Botany
M. S. S. N. Vaidya Vanita Maha Vidyalaya

R. Kalyani

2. Department of Micro-Biology

Best practices:

- Hands on experience for students in collecting various samples.

Expected outcomes:

- Analyze the usage of various equipments used in lab during collection and preservation of samples.
- Explore and apply their practical knowledge in their future endeavours.

Supporting documents / Photographs:



3. Department of Zoology

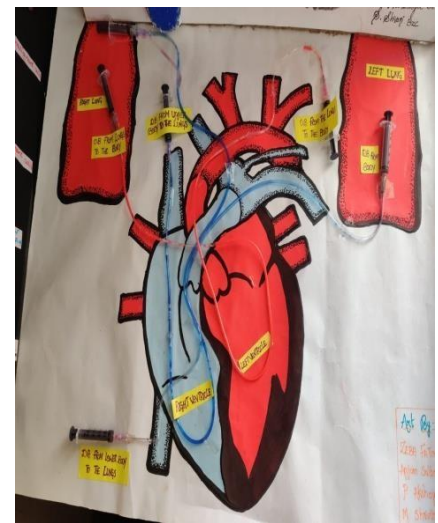
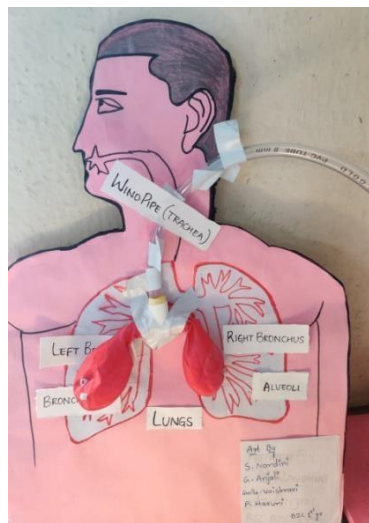
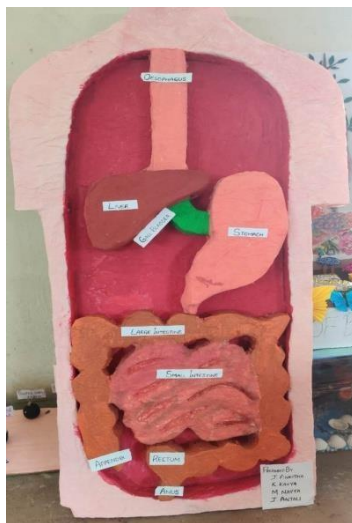
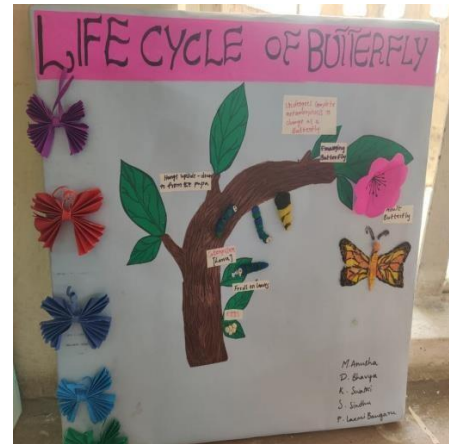
Best Practice: VERMICOMPOST PIT

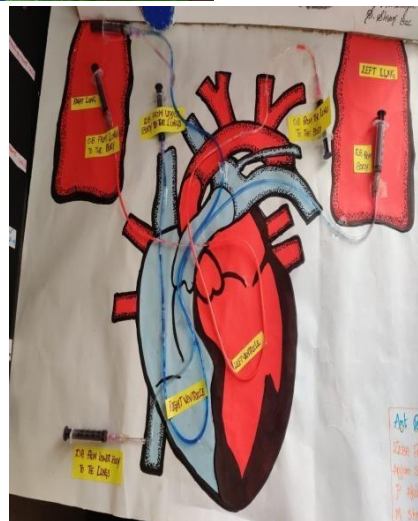
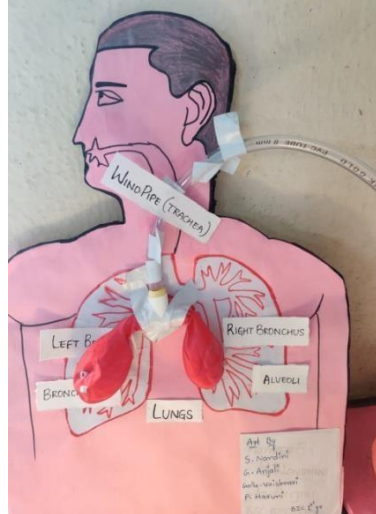
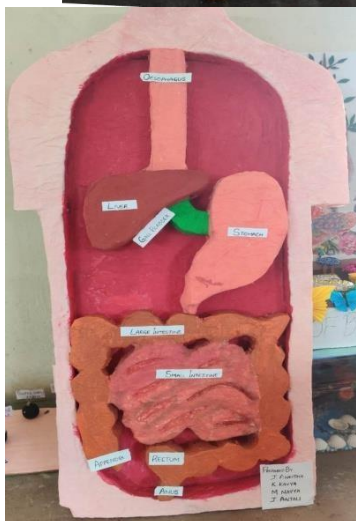
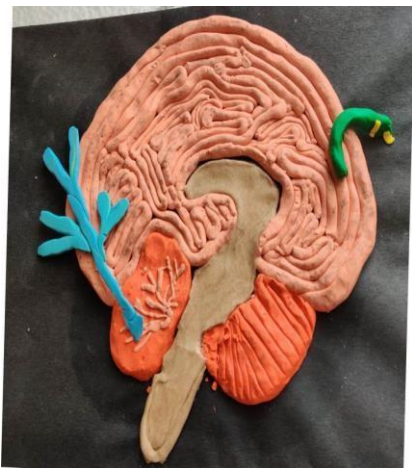
4. Vermicomposting is a natural process where by earth worm convert waste material with rigid structures into compost.
5. Vermicompost contains water soluble nutrients and is an excellent, nutrient –rich organic fertilizer and soil conditioner.
6. It is used in gardening and sustainable, organic farming.
7. We use dry waste for Composting (Dry cow dung, Hay and dry leaves) The compost is used for manuring the plants in the college garden.
8. Students are trained in preparing the organic compost which can be done at their homes too by utilizing the dry and wet waste (kitchen waste)

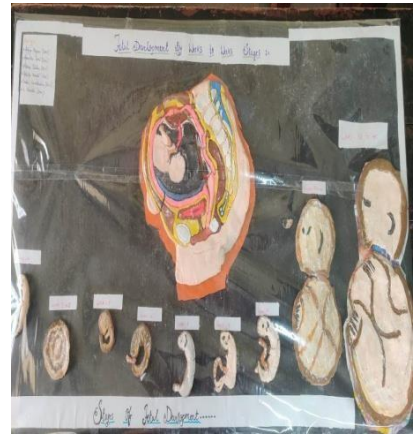
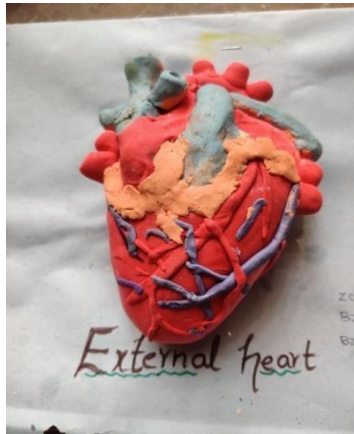


Beast Practice: Zoo club

Low cost working models using eco-friendly material are made by the students which are donated to the government schools where we conduct outreach programs. These models are used as aids for explaining the subject.







5. Department of Mathematics

Best Practice: Training in Quantitative Aptitude and Reasoning

Many students will write competitive examinations like **State Services, Civil Services, and Banking Service examinations**. The Department gives coaching to the interested students before and after college hours. It will save the money and time for the students as it will be provided free of cost and in the college

SAROJINI NAIDU VANITA MAHA VIDYALAYA

DEPARTMENT OF MATHEMATICS

Syllabus for Add on Course 2021 - 2022

Quantitative Aptitude

Duration (30 Hours)

I. Percentages

Basic concepts, comparing Two percentage.

Successive percentage changes.

Product constant Ratio, Fractions, Ratios.

Type of problems 1

Type of problems 2,

Tricks and shortcuts.

II Ratio & Proportions

Ratio and proportion definition.

1st proportion, 2nd proportion, 3rd proportion, 4th proportion &

Mean proportion

Types of problems, Tricks and Shortcuts.

III. Profit & Loss

Short cuts and Tricks for the problems on profit and Loss & Discount.

Out come:- Gain good knowledge in Traditional Methods in QA

and learn Short Cut Tricks or competitive exams.

6. Department of English

Best Practice: The Department will make the students to enact the poetry and dramas on stage. It will make them understand the pronunciation and the attire referred in the lessons. Bangle Sellers by Smt. Saorojini Naidu and Girish Karnad play 'Nagamandala' English students are few plays to quote



7. DEPARTMENT OF COMMERCE

1. **Best Practices** : a) Commerce Club
b) Departmental Library

2. **Expected Outcome** :
- a) Commerce club provides opportunity for the students to show case their talent and skills and it helps in their holistic development.
 - b) Departmental library comprises of 521 books that are issued to the students. It helps both the advanced learners and slow learners to improve their academic performance. The faculty donates the specimen copies received from the publishers to the Departmental library.
Students are issued reference and the books prescribed by the affiliated university. Students are allowed to use the books till the completion of the end semester examination.

3. **Activities undertaken**

- Under the best practices : a) Commerce Club conducted
- a) Commerce Antakshari on 28-2-19
 - b) Industrial Logo competition on 2-6-2022
 - c) Quiz competition on 7-11-2022
 - d) JAM session on 3-4-2023
 - e) Industrial Logo competition on 29-8-2023
 - f) Diwali Mela on 9-11-2023

- b) Under department library books were issued to the students

4. **Supporting documents/
Photographs** : Attached

Students participating in JAM session



STUDENTS PARTICIPATING IN LOGO COMPETITION



DIWALI MELA



RANGOLI COMPETITION



8. Department of Computer Science

Best Practices: Mobile as Computer

Many students cannot afford desktops or Lap tops, Whereas most of the students have smart phones. Department various suggests students to down load various Apps which will work as compilers, Web designers, AV editors. Using these tools students can run programs. Most of the Lab exercise can be done on mobile. This gives level playing field to economically week students.

- Compiler App
- Google Colab
- W3schools
- HTML Editor

3. Expected Outcome: Students can gain programming skills and get practical knowledge.

4. Activities Undertaken under the best practices: Google Colab (Python)

5. Supporting Documents and Photographs

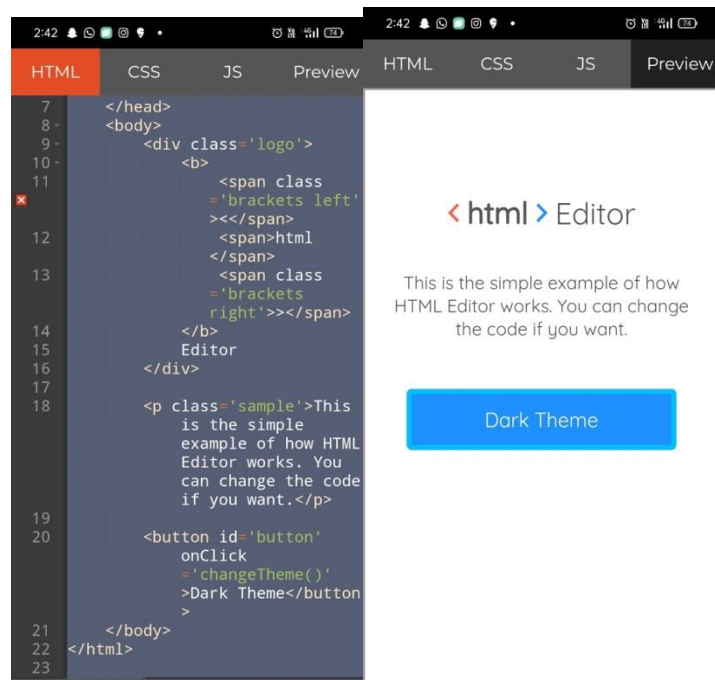
Google Colab



```
print("Hello, World!")
list=[1,2,3,5]
print(list)
```

Hello, World!
[1, 2, 3, 5]

HTML EDITOR



```
</head>
<body>
  <div class='logo'>
    <b>
      <span class
        ='brackets left'
      ><</span>
      <span>html
      </span>
      <span class
        ='brackets
          right'>></span>
    </b>
    Editor
  </div>
  <p class='sample'>This
    is the simple
    example of how HTML
    Editor works. You
    can change the code
    if you want.</p>
  <button id='button'
    onClick
      ='changeTheme()'
    >Dark Theme</button
  >
</body>
</html>
```

html Editor

This is the simple example of how HTML Editor works. You can change the code if you want.

Dark Theme

9. Department of Physics & Electronics Best Practices

1. Book Bank : Department maintains **Book Bank** which is repository of books generously donated/contributed by former students, staff and specimen copies of text books given by publishers . Students can barrow the essential books and materials required for their courses . This practice ensures that all students have access to the resources they need to succeed academically, regardless of financial constraints. This initiative promotes equity and inclusivity fostering a supportive learning environment where every student has the opportunity to excel



2. Optimal utilisation of equipment: The department prioritize the design of **Experiment Cycles** to maintain a better student – to – equipment ratio. This practice ensures that each student has ample access to apparatus /equipment, facilitating hands-on learning experience.

B.Sc (PHYSICS) PRACTICALS, SEM-IV

CYCLE-I

Date: 21-1-2024

GROUP	Roll No.	Strength	Experiment No.
MPCs	1175-22-468-001 TO -009	8	1
	1175-22-468-010 TO -018	8	2
	1175-22-468-019 TO -026	8	3
	1175-22-468-027 TO -035	9	4
	1175-22-468-036 TO -044	8	1
	1175-22-468-045 TO -053	8	2
	1175-22-468-054 TO -061	8	3
MPCs &MPC	1175-22-468-062 TO -068 & 1175-22-441-001 TO -004	9	4

Experiment No.	Name of the Experiment
1	Wedge method
2	Dispersive power of a Prism
3	Pulfrich refractometer
4	Sonometer

Anurachan
21/1/2024

B.Sc (PHYSICS) PRACTICALS, SEM-II

CYCLE-I

GROUP	Roll No.	Strength	Experiment No.
MPCs	1175-23-468-001 TO -009	9	1
	1175-23-468-010 TO -018	9	2
	1175-23-468-019 TO -027	9	3
MPCs	1175-23-468-028 TO -036	9	1
	1175-23-468-037 TO -045	9	2
	1175-23-468-046 TO -055	10	3

Experiment No.	Name of the Experiment
1	Lee's method
2	Electrical Kettle
3	Mechanical to Heat conversion

Anuradha
21/11/24